



# Enhanced Performance of Nanowire-Based All-TiO<sub>2</sub> Solar Cells using Subnanometer-Thick Atomic Layer Deposited ZnO Embedded Layer



Amir Ghobadi<sup>a,b</sup>, Halil I. Yavuz<sup>c,d</sup>, T. Gamze Ulusoy<sup>b,e</sup>, K. Cagatay Icli<sup>c,d</sup>, Macit Ozenbas<sup>c,d</sup>, Ali K. Okyay<sup>a,b,e,\*</sup>

<sup>a</sup> Department of Electrical and Electronics Engineering, Bilkent University, Ankara 06800, Turkey

<sup>b</sup> UNAM–National Nanotechnology Research Center, Bilkent University, Ankara 06800, Turkey

<sup>c</sup> Dept. of Metallurgical and Materials Engineering, Middle East Technical University, Ankara 06800, Turkey

<sup>d</sup> Center for Solar Energy Research and Applications (GUNAM), Middle East Technical University, Ankara 06800, Turkey

<sup>e</sup> Institute of Materials Science and Nanotechnology, Bilkent University, Ankara 06800, Turkey

## ARTICLE INFO

### Article history:

Received 6 November 2014

Received in revised form 10 January 2015

Accepted 18 January 2015

Available online 20 January 2015

### Keywords:

solar cells  
photovoltaic  
photoelectrochemical  
interface engineering  
titanium dioxide  
atomic layer deposition

## ABSTRACT

In this paper, the effect of angstrom-thick atomic layer deposited (ALD) ZnO embedded layer on photovoltaic (PV) performance of Nanowire-Based All-TiO<sub>2</sub> solar cells has been systematically investigated. Our results indicate that by varying the thickness of ZnO layer the efficiency of the solar cell can be significantly changed. It is shown that the efficiency has its maximum for optimal thickness of 1 ALD cycle in which this ultrathin ZnO layer improves device performance through passivation of surface traps without hampering injection efficiency of photogenerated electrons. The mechanisms contributing to this unprecedented change in PV performance of the cell have been scrutinized and discussed.

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## 1. Introduction

The projected growth of the silicon photovoltaic (PV) industry has been limited due to the material and manufacturing costs. Over the past decades, considerable studies were focused on finding alternative PV technologies which can offer low cost, mature processing technology together with high efficiencies. TiO<sub>2</sub> nanowire (NW) template-based hybrid solar cell structures are one of these alternatives because of its excellent optoelectronic and chemical properties in addition to providing high specific surface area, better electron transport and ability to strongly scatter light [1]. The n-type high band gap metal oxide such as TiO<sub>2</sub> NWs array has been extensively utilized as the electron transport layer in a wide range of solar cells such as hybrid solar cells, dye-sensitized solar cells [2] and organic solar cells [3]. Recently, in order to provide better absorption of light over the whole solar spectrum, different kind of semiconductors with higher absorption coefficients such as CdS [4,5], CdSe [6,7], CdTe [8–10], PbS [11] have

been used to sensitize metal oxide anode in solid/liquid state quantum dot/semiconductor sensitized solar cells. However, the functionality of semiconductor interfaces plays a crucial role in all hybrid solar cells in which trapping or recombination of charge carriers can reduce PV efficiency. In addition to photovoltaic applications, these interfaces have a great influence on the performance of photoelectrochemical and photocatalytic applications such as water splitting in which a better charge separation, transport and collection can be provided through the engineering of the semiconductors surface [12–14]. Controlling the impact of surface or interface-derived electronic states is, therefore, a prime goal in modern semiconductor processing. To this end, utilization of an interfacial semiconductor, typically a metal oxide with high energy band gap, layer is commonly employed [8,15,16]. However, the main drawback associated with such an interfacial layer is the fact that just a couple of nanometers of such a layer can significantly hamper injection efficiency. Therefore, an ultrathin homogeneous coating around the whole surface of NW is required. Among numerous methods available for the deposition of this passivation layer, atomic layer deposition (ALD) is able to coat pinhole-free metal-oxide films with angstrom-scale thickness.

Herein, we demonstrate that ALD coated ZnO embedded layer can efficiently passivate the NWs surface. Although the bare

\* Corresponding author at: Department of Electrical and Electronics Engineering, Bilkent University, Ankara 06800, Turkey.

E-mail address: [aokyay@ee.bilkent.edu.tr](mailto:aokyay@ee.bilkent.edu.tr) (A.K. Okyay).

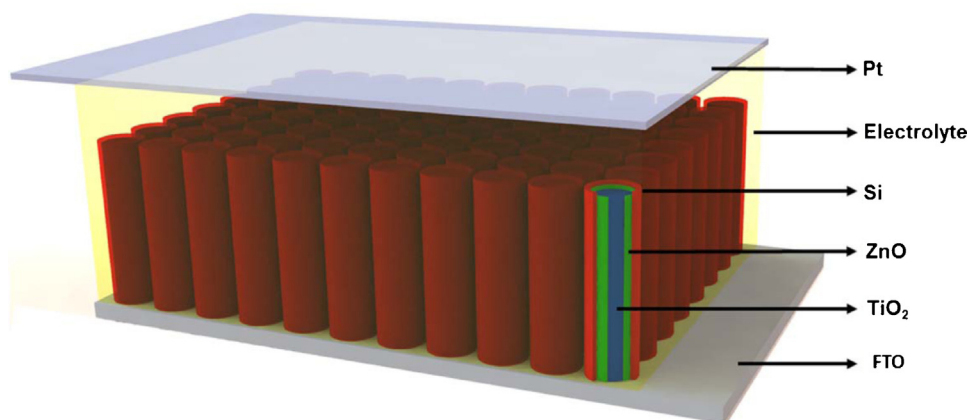


Fig. 1. (a) The 3D schematic of proposed  $\alpha$ -Si/ZnO/TiO<sub>2</sub> HJ solar cell.

structure (no ZnO interfacial layer) shows poor efficiency, device performance is boosted remarkably using the ZnO interfacial layer. With this in mind, the thickness of this interfacial layer plays a crucial role in solar cell performance, in that thicker layers of such a

high band gap material impede the electron injection noticeably. It is demonstrated that an optimized ultrathin layer paves the way to efficient devices by reducing recombination at the interface without hampering electron injection capability. It should be

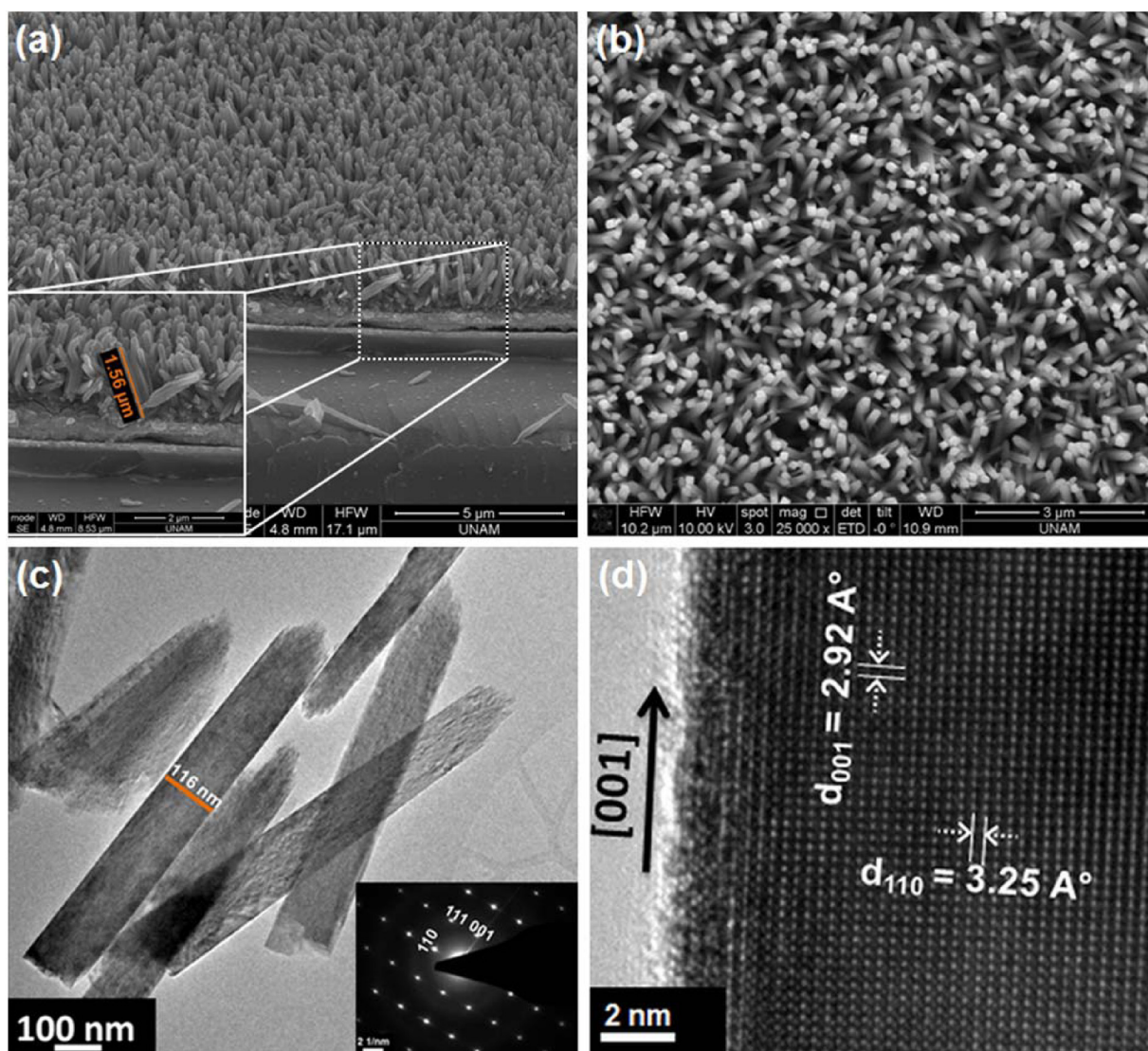


Fig. 2. SEM images of the NWs (a) cross section and (b) top view. (c) TEM image of ZnO coated TiO<sub>2</sub> NWs. Inset shows the SAED pattern of the sample. (d) HR-TEM image of the NWs showing the direction of growth and lattice spacing.

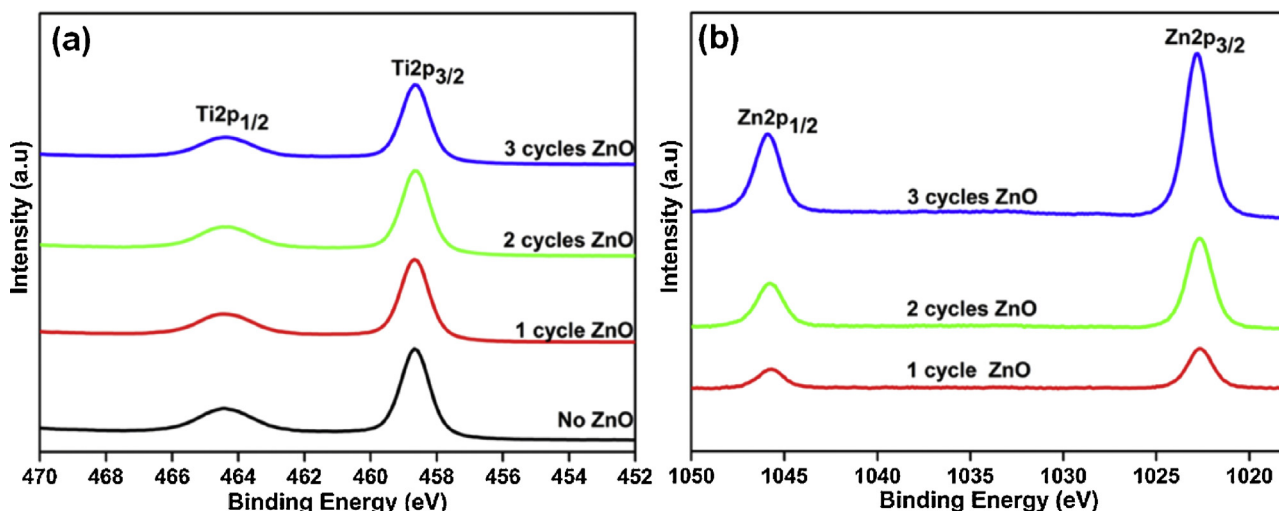


Fig. 3. XPS spectra of (a) Ti2p and (b) Zn2p are taken on the surface of TiO<sub>2</sub>-ZnO core-shell NWs.

noted that this modification on TiO<sub>2</sub> photoanode surface will improve device performance regardless of the material used as the absorbing layer. It will be discussed that the obtained results are not only promising for PV technology but also they can provide a substantial improvement in the future performance-enhanced photoelectrochemical and photocatalytic cells such as water splitting in which an ultrathin shell layer is desirable for minimizing minority carriers diffusion path while improving charge separation and reducing the recombination rates at the interface. In order to prove the effectiveness of this idea in the photovoltaic performance enhancement, a thin layer of amorphous Si ( $\alpha$ -Si) is used as absorbing layer on TiO<sub>2</sub> NWs and liquid electrolyte as a hole conductor layer. The impact of ZnO interfacial layer thickness on PV parameters such as open circuit voltage and short circuit current is investigated and discussed.

## 2. Experimental procedure

### 2.1. Chemicals

Ethanol, acetone, titanium butoxide (Ti(OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>)<sub>4</sub>) (97%) and hydrochloric acid HCl (36%) are all from Sigma-Aldrich

Co and used as received. FTO coated glass (7  $\Omega$  sq<sup>-1</sup>), Iodalyte HI electrolyte are all purchased from Solaronix.

### 2.2. Growth of TiO<sub>2</sub> nanowire array

We prepared the uniform TiO<sub>2</sub> NWs on FTO coated glass by hydrothermal technique according to our previous report [17,18]. Briefly, HCl (20 ml) and DI (20 ml) are mixed in a teflon-lined stainless steel autoclave (45 ml) for 10 min and afterward 0.8 ml titanium butoxide is added. After mixing with precursor for 30 min, FTO is immersed into solution and kept at 140 °C for 4 hours.

### 2.3. Deposition of ZnO and Si layers on TiO<sub>2</sub> nanowires

NWs are coated with ZnO by ALD reactor. The substrate temperature is kept at 250 °C during the process. Various samples are prepared with different number of cycles (1, 2 and 3) of ZnO as an ultrathin layer on TiO<sub>2</sub> NWs. For ZnO deposition by ALD, diethylzinc ((C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>Zn or DEZn, Sigma-Aldrich) and HPLC-grade water (H<sub>2</sub>O) are used as the zinc and oxygen precursors, respectively. A thin layer of p-type  $\alpha$ -Si is deposited on prepared

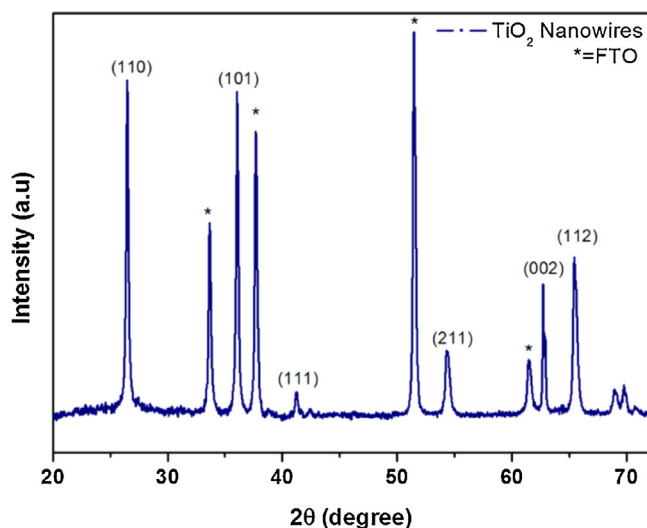


Fig. 4. XRD patterns of ZnO coated TiO<sub>2</sub> NWs

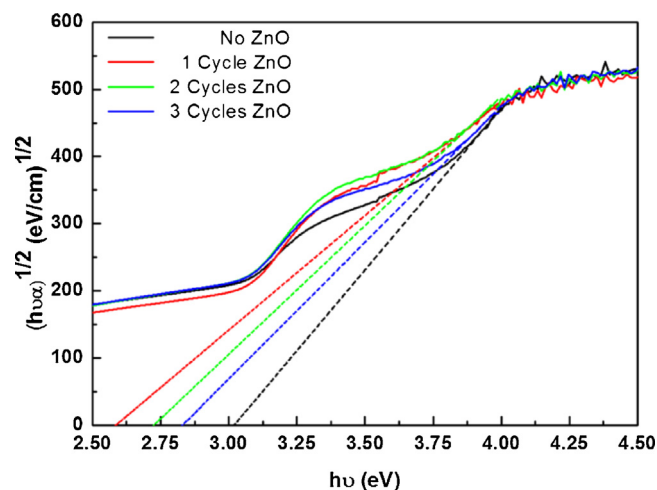


Fig. 5.  $(\alpha h\nu)^{1/2}$  versus  $h\nu$  plot for TiO<sub>2</sub>/ZnO core-shell heterostructures with different ZnO cycles.



**Table 1**  
Effective optical band gap of TiO<sub>2</sub>/ZnO core-shell heterostructures.

# of ZnO ALD cycles	Band gap (eV)
0	3.02
1	2.58
2	2.72
3	2.83

TiO<sub>2</sub>/ZnO core-shell NWs by RF magnetron sputter deposition system. The deposition of the  $\alpha$ -Si layer is carried out under  $10^{-3}$  Torr Ar gas pressure employing 100 W RF power with a deposition rate of 0.3 Å/s for 10 mins.

#### 2.4. Materials characterization

Scanning electron microscope (SEM, FEI–Quanta 200 FEG) and transmission electron microscope (TEM, FEI model Tecnai G2 F30) are used to investigate the morphology and dimensions of the NWs. The phase structures of the obtained samples were identified using a Pananalytical (X'pert Pro MPD) instrument and XRD patterns were collected over the  $2\theta$  angular range of 20–70° using Bragg–Brentano geometry. High resolution X-Ray photoelectron spectroscopy (XPS) measurement is performed to verify the existence of ultrathin ZnO layer.

#### 2.5. Device Assembly

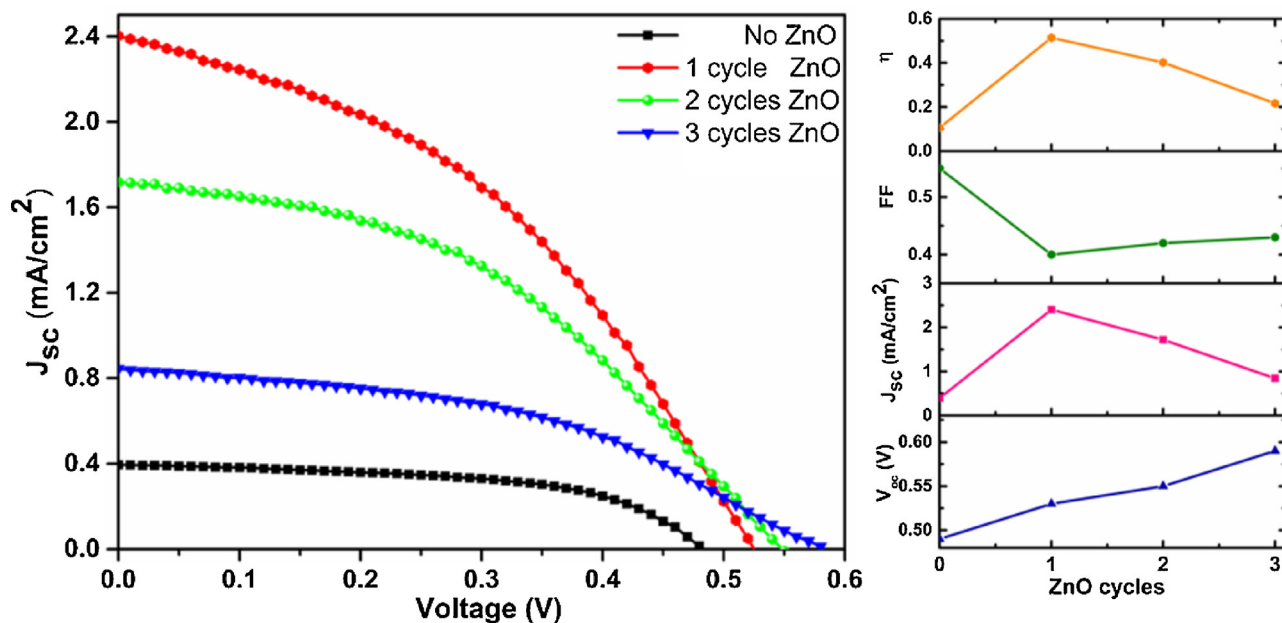
3D schematic of the  $\alpha$ -Si/ZnO/TiO<sub>2</sub> heterojunction (HJ) solar cell structure is shown in Fig. 1. The proposed structure consists of 5 main parts; 1) hydrothermally grown TiO<sub>2</sub> NWs on FTO coated glass, 2) ALD deposited ZnO layer coated on TiO<sub>2</sub> NWs, 3) a thin layer of  $\alpha$ -Si deposited by RF magnetron sputtering on TiO<sub>2</sub>/ZnO HJ as absorbing layer, 4) redox electrolyte iodide/triiodide based electrolyte ( $I^-/I_3^-$ ) as a hole transfer mediator and finally 5) Pt-coated conducting glass as a counter electrode. For device assembly, the Si-coated TiO<sub>2</sub>/ZnO photoanode and platinum counter electrode are sandwiched together using a cell holder. The internal space of device is filled with 1 ml syringe electrolyte through the backfilling technique

(see Fig. 1). The area of the electrode is controlled using a mask of 0.20 cm<sup>2</sup> area.

### 3. Results and discussion

Fig. 2 shows SEM and TEM images of fabricated ZnO coated NWs. Fig. 2 (a–c) illustrates that densely-packed NWs with an average length of 0.9–1.6  $\mu$ m have a diameter in the order of 80–150 nm. The selected area electron diffraction (SAED) pattern and high resolution TEM (HRTEM) images provide further confirmation on single crystalline phase of each individual TiO<sub>2</sub> NW. According to the HRTEM image, the lattice fringe spacing for the sample is 3.25 Å which corresponds to the interspacing of the family of the (1 1 0) planes of tetragonal rutile phase of TiO<sub>2</sub> and shows a direction growth occurred along the [001]. In addition to this, high resolution XPS measurement is performed to verify the existence of ultrathin ZnO layer (it is difficult to distinguish this angstrom-scale shell layer from TiO<sub>2</sub> NW core in TEM image). The spectrum of TiO<sub>2</sub> and ZnO peaks including Ti2p<sup>3/2</sup>, Ti2p<sup>1/2</sup> and Zn2p<sup>3/2</sup>, Zn2p<sup>1/2</sup> peaks are observed at 458.61 eV, 464.26 eV and 1022.18 eV and 1045.24 (as depicted in Fig. 3(a,b)) which is in line with previous reports [19–21]. Finally, to explore the orientation of the NWs growth and their crystalline structure, XRD analysis is carried out on the obtained samples. The XRD pattern for the bare TiO<sub>2</sub> NWs array is depicted in Fig. 4. All diffraction peaks are attributed to the FTO and rutile phase of TiO<sub>2</sub>. According to data shown here, the diffraction peaks pattern is in agreement with rutile phase of TiO<sub>2</sub> with a dominant peak located at 36.25° belongs to NWs grown along (101) direction. Since the shell layer is ultrathin, we observed no peak related to ZnO.

As the first step in the optical characterization of the NWs, using transmission data obtained from UV-VIS-NIR spectrophotometer, optical band gap,  $E_g$ , is experimentally determined by extrapolating the linear portion of the Kubelka–Munk function,  $(\alpha h\nu)^{1/2}$ , versus photon energy,  $h\nu$ , graph shown in Fig. 5, where  $\alpha$  is the absorption coefficient. Table 1 presents average optical band gap values calculated from the above. While the optical band gap of the bare rutile TiO<sub>2</sub> NW arrays is found to be 3.02 eV, that for TiO<sub>2</sub>/ZnO core-shell heterostructures with 1 cycle ZnO, effective optical band gap is reduced by 15% down to 2.58 eV. However, for thicker shell



**Fig. 6.** (a) J–V characteristics and (b) Photovoltaic characteristics of solar cells for different ZnO cycles.

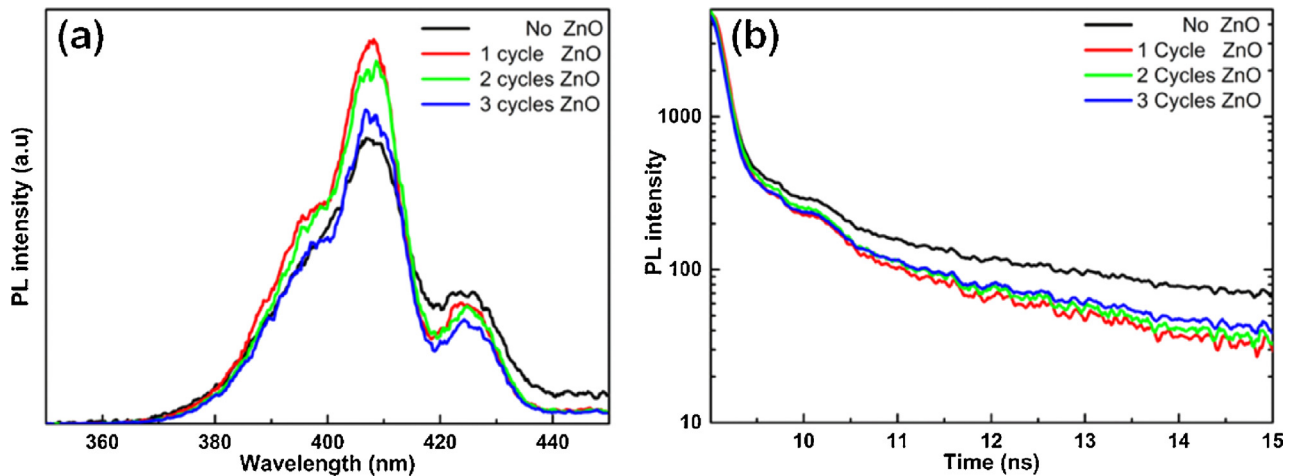


Fig. 7. Room temperature (a) PL spectra, (b) TRPL spectra for different ALD ZnO cycles coated samples

layers a gradual increase in band gap is followed in which for 3 cycles coated ZnO shell layer effective band gap reaches to 2.83 eV. This reduction in the band gap for first ALD cycles can be attributed to quantum confinement effect in the ultrathin ZnO shell. In upcoming sections it will be shown that band alignment of  $\text{TiO}_2/\text{ZnO}$  interface is similar to type-II band alignment. Therefore, for ultrathin shells, a significantly reduced effective band gap results for excitonic transitions between valance band of  $\text{TiO}_2$  core and conduction band of ZnO shell layer. This reduced effective band gap is lower than those of both pure  $\text{TiO}_2$  and ZnO. Moreover, this band gap narrowing and extension of the absorption edge toward higher wavelengths enable visible light driven photocatalytic and photoelectrochemical water splitting which is a hot research topic in recent years [14,22–24].

In order to investigate the impact of ZnO interfacial layer in the device PV performance, current-voltage ( $J$ - $V$ ) characteristics of different cells are measured by a Keithley 2440 source meter (AM1.5 G,  $100 \text{ mW}/\text{cm}^2$ ). As it can be observed in Fig. 6(a), the number of ZnO cycles deposited on the  $\text{TiO}_2$  has a distinct effect on the performance of the solar cells. In the case of 1 ALD ZnO cycle,  $\alpha\text{-Si}/\text{ZnO}/\text{TiO}_2$  HJ solar cell device efficiency makes a steep rise to a maximum efficiency of 0.514%, which is nearly a five-fold increase compared to nominal efficiency of 0.106% for bare device (no ZnO interfacial layer). After the first cycle onwards, the efficiency starts to follow a decreasing trend down to 0.216% for 3 cycles. The trend associated to each critical PV parameter is shown in Fig. 6(b).

To scrutinize the physics behind this unprecedented change in device performance, a closer investigation of short circuit current ( $J_{sc}$ ) and open circuit voltage ( $V_{oc}$ ) trend is required. As it can be clearly seen, the  $J_{sc}$  follows the same trend as efficiency ( $\eta$ ) while  $V_{oc}$  shows a monotonically increasing trend from 0.49 V for the bare sample to 0.59 V for 3 ALD cycles of ZnO. Since the deposition of a sub-nanometer ZnO layer cannot change absorption considerably, this substantial improvement in PV parameters of the device can be attributed to reducing the loss mechanisms at the interface through reduction in density of trap states or retardation of recombination kinetics in the HJ interface.

According to the NWs hydrothermal growth method, insufficient Ti oxidation can induce surface dangling bonds such as oxygen vacancies or Ti interstitials which reduce the charge collection efficiency in the PV device. On the other hand, because of the self-limiting nature of the ALD technique, ZnO layer is expected to have few zinc interstitials and oxygen vacancies and in the meantime it passivates the traps on the  $\text{TiO}_2$  surface [25]. This is further investigated using photoluminescence (PL) spectroscopy since a correlation between the PL intensity and the defect densities are expected. Fig. 7(a) depicts the room temperature PL spectra of the bare and ZnO-coated samples for an excitation wavelength of 320 nm. A near-band-edge emission (NBE) at 408 nm and a shallow trap emission (STE) centered at 424 nm can be shown for all samples which is consistent with previous reports [26]. Following the deposition of only a single ALD cycle of

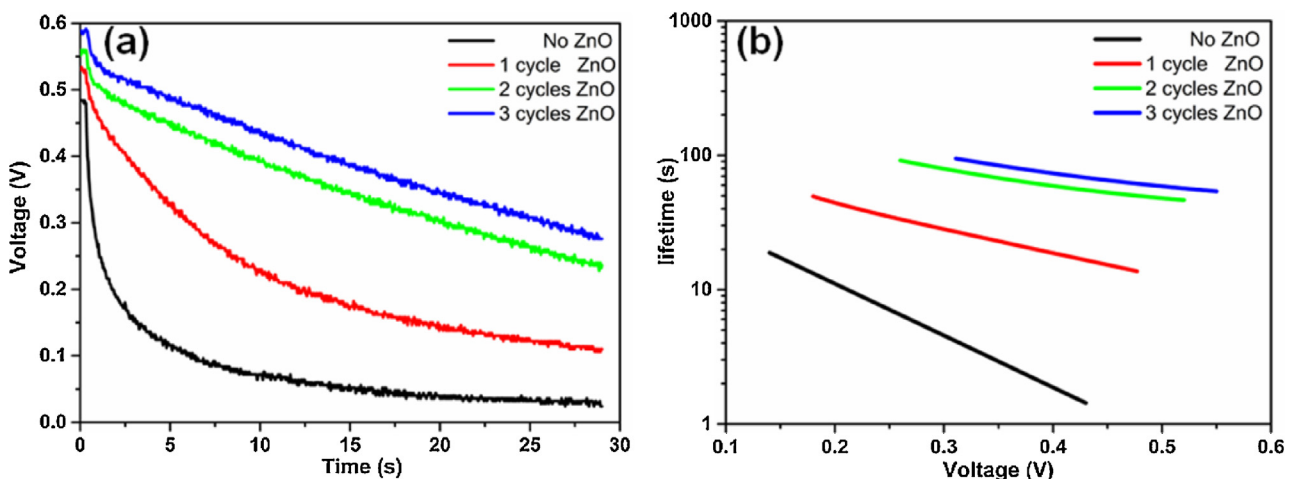


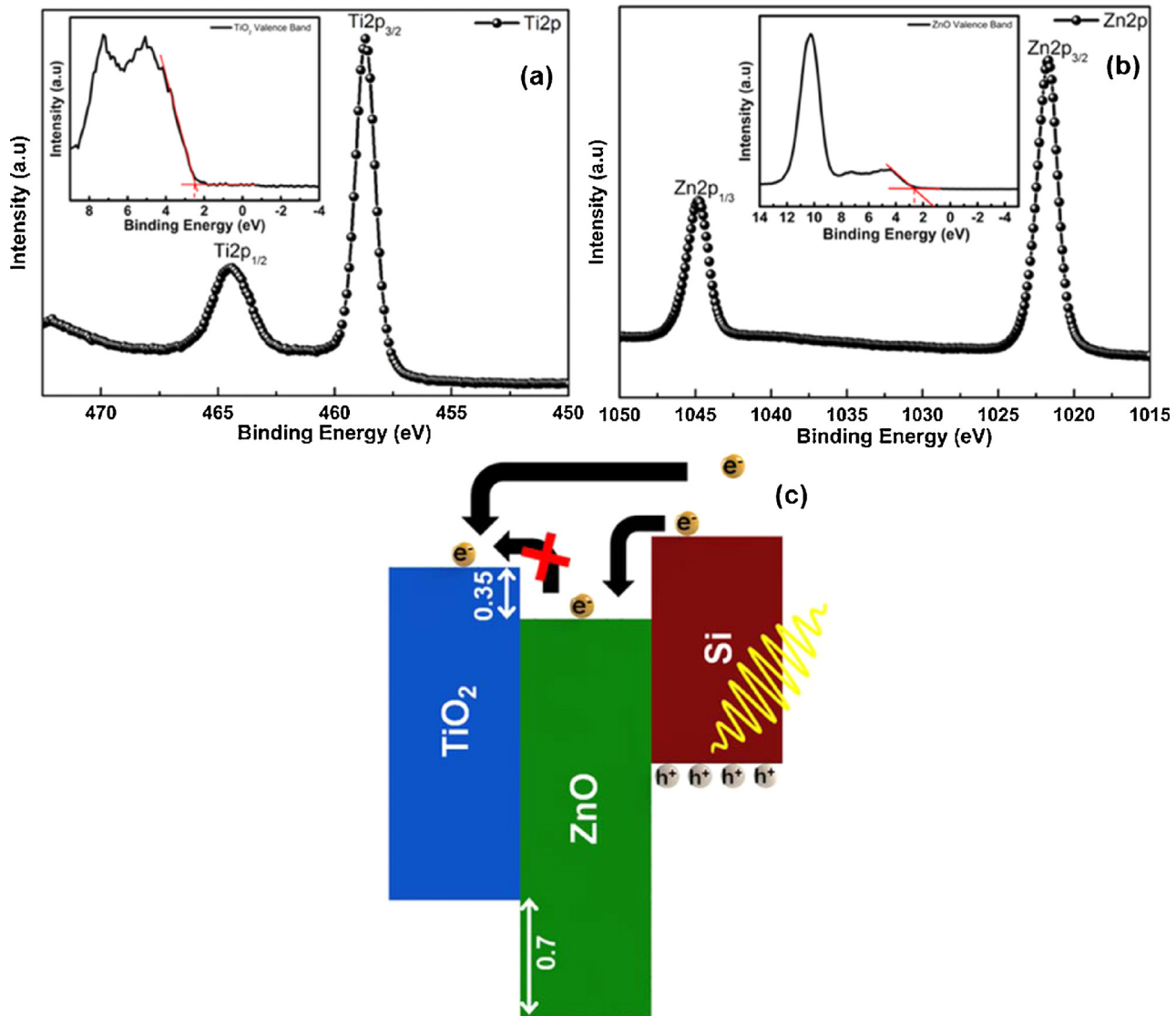
Fig. 8. (a) Experimental  $V_{oc}$  decay and (b) determined carrier lifetimes for different ALD ZnO cycles coated samples

ZnO layer, NBE increases considerably while STE is quenched compared to that of bare  $\text{TiO}_2$  NWs. For the next ALD cycles both peaks exhibit decreasing trends. These results reveal that the first ALD ZnO cycle will effectively terminate surface trap states on the NWs surface and suppress the activity of non-radiative recombination sites which in turn increases the NBE intensity together with quenching the STE [27,28]. To get further insight into the carrier dynamics, the room temperature time-resolved photoluminescence (TRPL) measurements for bare and passivated samples are conducted (Fig. 7(b)). As it is expected, all coated samples show faster decays than the bare one which is in agreement with previous results which reveals the passivation of surface traps via coating with ZnO layer. It is worthy to note that by moving to thicker layers, the photoexcited carriers show more long-lived behavior. In upcoming sections, it will be shown that energy bands across  $\text{TiO}_2/\text{ZnO}$  interface are aligned in a way that an efficient charge separation can be provided in which photo excited electrons will diffuse to the shell layer while holes will be confined in the core. These results can also explain the reduction in NBE intensity for thick ZnO layers in which this charge separation at the HJ interface reduces the probability of band-to-band recombination and consequently lessens NBE intensity.

To probe recombination kinetics in complete PV device,  $V_{oc}$  decay, as a powerful measurement tool to estimate electron lifetime, is utilized. For this, we monitored the transient  $V_{oc}$  as a function of time upon switching off the light to investigate the electron recombination kinetics. As it can be clearly seen in Fig. 8(a), HJ solar cell with no interfacial ZnO layer exhibits a sharp decay. However, the  $V_{oc}$  of the ZnO passivated solar cells demonstrates much slower decay rates which imply slower recombination kinetics and longer electron lifetime for the excited electrons. The decay rate can be directly related to the electron lifetime by Eq. (1).

$$\tau_n = -\frac{kT}{e} \left( \frac{dV_{oc}}{dt} \right)^{-1} \quad (1)$$

The charge lifetimes determined from the  $V_{oc}$  decay measurements are shown in Fig. 8(b). From this, it is clear that by deposition of just one cycle of ZnO layer charge lifetime has increased and this enhancement is continued for two cycle coated ZnO layer but after this point results are almost similar for two and three cycles. These results prove that, together with passivation of surface traps, this ultrathin ZnO embedded layer can also reduce recombination rate via providing an efficient charge separation at



**Fig. 9.** CLs and VB spectra of (a) Zn  $2p_{3/2}$  and (b) Ti  $2p_{3/2}$  recorded on pure ZnO and  $\text{TiO}_2$  samples (VBM values are determined by extrapolating of leading edge to the base line.) (c) experimentally determined energy band alignment for the HJ.

**Table 2**

Results obtained from XPS valence band spectra

Sample	Region	Binding Energy (eV)
TiO <sub>2</sub>	Ti 2p <sup>3/2</sup>	458.68
	VBM	2.59
ZnO	Zn 2p <sup>3/2</sup>	1021.58
	VBM	2.62
TiO <sub>2</sub> /ZnO	Zn 2p <sup>3/2</sup>	1022.18
	Ti 2p <sup>3/2</sup>	458.61

the interface. Despite of its excellent PV characteristics, this core-shell structure has a great potential for photoelectrochemical water splitting due to having low recombination rates, better carrier separation and high collection capability which are prominent factors for hydrogen production rate of water splitting cell [29–32].

As shown earlier, deposition of one cycle ZnO embedded layer boosts power conversion efficiency up to its maximum and moving toward thicker layers, efficiency gradually falls down. In order to explore the physics behind this drop in device performance, understanding the electron injection mechanism at HJ interface is an imperative task. Although some reports show a band alignment where the position of conduction band of the ZnO layer is at higher energy than that of TiO<sub>2</sub> [33–35], some other reports claim the opposite [36,37]. For this aim, an analysis technique of Kraut [38] based on high resolution XPS measurement is adopted to provide band alignment in TiO<sub>2</sub>/ZnO heterostructure via estimating valance and conduction band offsets ( $\Delta E_V$  and  $\Delta E_C$ ). Firstly, the energy difference between Zn2p and Ti2p core levels ( $\Delta E_{CL}$ ) in the TiO<sub>2</sub>/ZnO core-shell sample is found to be 563.57 eV Fig. 3. Also, the difference between core level energy and valance band maximum, ( $\Delta E_{CL} - \Delta E_{VBM}$ ) for both TiO<sub>2</sub> NWs sample and ZnO thin film layer is calculated as it is illustrated in Fig. 9(a,b). Table 2 summarizes all obtained data. Moreover, the optical band gap for TiO<sub>2</sub> NW array has been estimated to be 3.02 eV using transmission data obtained from UV-VIS-NIR spectrophotometer and extrapolating the linear portion of the Kubelka–Munk function, see Fig. 5. For this analysis, the band gap for ZnO is taken as 3.37 eV from our previous work [39]. Finally, employing Eqs. (2) and (3) the amounts of  $\Delta E_V$  and  $\Delta E_C$  are calculated to be 0.7 eV and 0.35 eV, respectively.

$$\Delta E_V = (E_{CL} - E_{VBM})_{TiO_2/NW} - (E_{CL} - E_{VBM})_{bulkZnO} + \Delta E_{CL} \quad (2)$$

$$\Delta E_C = (E_{gZnO} - E_{gTiO_2} - \Delta E_V)_{TiO_2/ZnO\text{core-shell}} \quad (3)$$

This analysis suggests a band alignment depicted in Fig. 9(c). As it can be seen from the schematic, conduction band offset in TiO<sub>2</sub>/ZnO interface is in a way that, a portion of photo generated electrons in the absorbing layer will be captured by ZnO quantum well while injecting into TiO<sub>2</sub> conduction band. This probability is intensified by increasing the width of ZnO quantum well which in turn leads to a reduction in the PV efficiency of the device.

#### 4. Conclusions

In summary, we demonstrated that using angstrom-thick atomic layer deposited ZnO shell layer can significantly enhance the power conversion efficiency of PV devices based on TiO<sub>2</sub> NWs. Such an ultrathin layer contributes to device performance enhancement via reducing the recombination mechanisms in the interface without significantly impeding the injection of photo-generated carriers due to its negligible thickness. This improvement, however, is independent of the type of absorbing layer ( $\alpha$ -Si is used here for proof-of-concept purpose). The results presented here are considered a

paradigm shift not only in NW-based PV technologies but also in other photoelectrochemical and photocatalytic applications and serve as a beacon for future performance enhanced NW-based all-TiO<sub>2</sub> solar cell and water splitting devices.

#### ACKNOWLEDGEMENTS

This work was supported by the Scientific and Technological Research Council of Turkey (TUBITAK), grant numbers 109E044,112M004, 112E052, and 113M815. A.K.O. acknowledges support from the Turkish Academy of Sciences Distinguished Young Scientist Award (TUBA GEBIP).

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