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Structural and electronic properties of MoS₂, WS₂, and WS₂/MoS₂ heterostructures encapsulated with hexagonal boron nitride monolayers

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In this study, we investigate the structural and electronic properties of MoS₂, WS₂, and WS₂/MoS₂ structures encapsulated within hexagonal boron nitride (h-BN) monolayers with first-principles calculations based on density functional theory by using the recently developed non-local van der Waals density functional (rvv10). We find that the heterostructures are thermodynamically stable with the interlayer distance ranging from 3.425 Å to 3.625 Å implying van der Waals type interaction between the layers. Except for the WS₂/h-BN heterostructure which exhibits direct band gap character with the value of 1.920 eV at the K point, all proposed heterostructures show indirect band gap behavior from the valence band maximum at the Γ point to the conduction band minimum at the K point with values varying from 0.907 eV to 1.710 eV. More importantly, it is found that h-BN is an excellent candidate for the protection of intrinsic properties of MoS₂, WS₂, and WS₂/MoS₂ structures. *Published by AIP Publishing*. [http://dx.doi.org/10.1063/1.4998522]

I. INTRODUCTION

The successful experimental realization of monolayer graphene leads to a new research field with vast amount of interest in two dimensional (2D) structures because of their fascinating properties. The extent of research on 2D crystals, beyond graphene, based on atomically thin films of layered semiconductors such as the family of transition metal dichalcogenides (TMDs) (MX₂, M = Mo, W; X = S, Se, Te) is emerging and has been increasing ever since. The attractiveness of these 2D materials is mainly due to their excellent electrical, mechanical, and optical properties which offer great potential for novel applications, for example, in devices like light emitting diodes (LEDs),¹ transistors,² sensors,³ and photodetectors.⁴ The MX₂ monolayers also provide a fundamental requirement for common electronic devices such as novel optoelectronic and photovoltaic applications due to the presence of direct band gap behavior.^{5–11} Moreover, the idea of exploration of a variety of heterostructures from 2D layered TMDs has driven a new platform for modern potential applications such as tunnelling transistors,¹⁶ memory devices,¹² and ultrathin photodetectors.¹³ Therefore, recently, special attention has been paid to these 2D crystals both theoretically and experimentally.^{5,14,15} Along these lines, abinitio calculations play a crucial role in introducing new possible 2D layered heterostructures by artificially stacking the monolayer of MX₂ for experimentalists and engineers to construct state-of-the-art electronic devices with less effort.

Among the TMDs, MoS_2 has been extensively studied,^{17–19} since its properties can be tailored by playing with its environment or even its thickness. For example, drastical changes in its band gap with respect to the adsorbates, strain, interlayer interaction, and charges in neighbouring dielectrics were reported.¹⁷⁻¹⁹ Owing to similar lattice constants, monolayer MoS₂ and WS₂ are naturally potential substrates to supporting each other. In contrast to planar graphene or monolayer hexagonal boron nitride (h-BN), monolayer MoS₂ or WS₂ is not single-atom thick since the hexagonally packed layer of Mo or W atoms is sandwiched between two layers of S atoms in their monolayer structures. Furthermore, the absence of dangling bonds on their surfaces and charge traps provides great opportunity to create high quality nanoelectronic devices and heterostructures. In their heterostructures, these single layers are mainly stacked together by van der Waals (vdW) interactions. MoS₂ and WS₂ have an indirect band gap in their bulk form, however, when they are thinned down to a single layer, the band gap becomes direct. In a recent study, it has been shown that monolayer MoS₂ strongly emits light because of the 1.8 eV direct band gap at the K point in the Brillouin zone.²⁰ This is also verified by experimental groups using a photoluminescence method.^{21–23} Similarly, the direct band gap of 1.9 eV is obtained for chemical vapor deposition (CVD) islands of monolayer WS₂.²⁴ Breaking of inversion symmetry is the reason for transformation to a direct band gap in the monolayer of TMDs. This has also been confirmed experimentally by circularly polarized light experiments that lead to valley polarization effects.²⁵⁻²⁷ TMDs also have large exciton binding energies and strong photoluminescence leading to novel material platforms for optoelectronic applications.

In order to offer functional materials with high performances, one needs to take advantage of 2D heterostructures. For example, a new generation of field-effect-transistors²⁸ with a high on-off ratio (>10³) and a current density of up to 5000 A cm⁻² has been fabricated using vertically stacked graphene–MoS₂–metal heterostructures. Recent studies have reported the successful growth of vertical heterostructures by chemical vapor deposition (CVD), for example, see

graphene/h-BN,²⁹ MoS₂/grapheme,³⁰ MoSe₂/grapheme,³¹ $MoS_2/h-BN$,³² $WS_2/h-BN$,³³ MX_2/SnS_2 (M = Mo, W; X = S, Se),³⁴ and WS2/MoS₂.³⁵ The CVD method is one of the best common techniques in order to grow single crystals at a large scale and in a controllable manner, which are very important to fabricating high-quality MoS₂ electronic devices. Because of this, researchers have devoted subsequent efforts to the CVD growth of MoS₂ on different substrates.^{37,38} One example is the growth of monolayer MoS₂ over h-BN by using the seeding method.³⁹ Furthermore, the direct growth of monolayer MoS₂ has been demonstrated on exfoliated highquality h-BN flakes without using seeding methods, which yield a clean interface between the grown MoS₂ and the h-BN substrate.^{39,40} Moreover, the fabrication cost of higher-quality devices is reduced in this method. The advantage of using h-BN as a substrate to grow 2D materials comes from its atomically thin structure without any dangling bonds and charged impurities within it. Furthermore, the properties like being an insulator and a chemically inert system make the h-BN monolayer an ideal substrate for protecting overgrown high quality TMDs for electronic applications and even for improving the device performance. Such examples, for instance, MoS₂ deposited over an h-BN substrate, are reported in the literature.^{17,36}

The main focus of this study is to characterise novel h-BN/MoS₂/h-BN, h-BN/WS₂/h-BN, and h-BN/WS₂/MoS₂/ h-BN heterostructures by performing extensive firstprinciples calculations. Our aim in these first-principles calculations is to control the magnitude of band gaps in the heterostructures to find a way for developing high-quality durable single- and hetero-layer TMD electronic devices such as optoelectronic devices and field effect transistors. The effects of intercalation of monolayer MoS₂, monolayer WS₂, and WS₂/MoS₂ heterolayer into bilayer h-BN and interlayer hybridisation on the structural and electronic properties of monolayer MoS₂, monolayer WS₂ and WS₂/MoS₂ heterolayer are investigated in detail using density functional theory (DFT) with a recently developed non-local van der Waals (vdW) density functional by Vydrov and Van Voorhis (rVV10).⁴¹ Our computational results suggest that the encapsulated heterostructures, WS₂/h-BN, MoS₂/h-BN, and WS₂/ MoS₂/h-BN, are energetically stable, and the interaction between the layers is weak due to the large interlayer distance, binding energies in the range of physical adsorption, and negligible buckling of the h-BN monolayer indicating van der Waals type interaction. The explanation for interlayer interactions in the encapsulated heterostructures is crucial for proposing and tailoring new technological applications. To the best of our knowledge, there have been no first-principles investigations on the effect of h-BN monolayer on the electronic structure and structural properties of the encapsulated h-BN/WS₂/MoS₂/h-BN structure. The band structure analysis shows that the direct band gap character of WS₂ is still persistent on the WS₂/h-BN heterostructure. However, the band gap becomes indirect for monolayer MoS₂ and WS₂/MoS₂ heterostructure when deposited onto or sandwiched with the h-BN monolayer. It is also worth mentioning that when monolayer MoS₂, monolayer WS₂ and WS₂/MoS₂ heterolayer are deposited onto monolayer h-BN and sandwiched between the h-BN bilayers, the band gap value of MoS_2 is reduced to the amount of 0.89 eV in the h-BN/MoS₂/h-BN heterostructure and 0.14 eV in the MoS₂/ h-BN structure. However, the band gap value of WS₂ is diminished to the amount of 20 and 710 meV in interfaced and sandwiched structures, respectively. For the WS₂/MoS₂ heterolayer structure, the indirect band gap is retained with the band gap values varying from 0.967 eV for WS₂/MoS₂/h-BN to 1.110 eV for h-BN/WS₂/MoS₂/h-BN. We further found that the atomically thin flat h-BN monolayers are buckled within the range of 0.01 Å and 0.18 Å. This buckling suggests that the h-BN monolayer can be used for gateinsulating materials with minimized interaction between monolayer h-BN and TMDs. Our theoretical investigations offer an improvement in the quality of monolayer MoS_2 , monolayer WS₂, and WS₂/MoS₂ heterolayer by using h-BN monolayer and provide band structure engineering of TMDs which is a very useful guide for the construction of TMD based nanoelectronics. From our theoretical predictions, we show that the h-BN monolayer eliminates problematic interfacial defects in TMDs, and can be chosen as an ideal candidate for a high-speed FET.

II. COMPUTATIONAL METHOD

All calculations are carried out within the framework of plane-wave density functional theory as embodied in the Quantum ESPRESSO package,⁴² including long-range dispersive interactions with the van der Waals interactioncorrected density functional (DFT/rVV10).⁴¹ The rVV10 is one of the most popular ones among the DFT + D, DFT + D2, and vdW-family in order to effectively describe the nonlocal and long-range nature of the vdW interaction in the layered materials. This is because it uses the electron density to directly obtain the dispersion interactions, however, DFT + D, DFT + D2, and vdW-family correct the DFT total energy by adding an empirical atom-pairwise interaction correction, parametrized by atomic C₆ coefficients. Therefore, this functional is called a non-local correlation functional in which the total exchange-correlation (xc) energy is defined as $E_{xc} = E_{xc}^0 + E_c^{nl}$, where E_{xc}^0 is the semilocal xc and E_c^{nl} is the nonlocal correlation energy discussed in detail in Ref. 41. The rVV10 method gives the most accurate results for intralayer and interlayer lattice constants for 28 layered materials which have been confirmed from experimental results in Ref. 43. The electron-ion interaction is described by ultrasoft pseudopotentials.⁴⁴ A plane-wave basis set with a kinetic energy cutoff of 60 Ryd is adopted to expand the single-particle Kohn-Sham orbitals. Brillouin zone sampling of electronic states is approximated by using the sets of special k-points corresponding to the $(36 \times 36 \times 1)$ Monkhorst-Park mesh for monolayer MoS₂, monolayer WS₂, and WS₂/MoS₂ heterolayer and $(12 \times 12 \times 1)$ Monkhorst-Park mesh for proposed heterostructures.⁴⁵ To hinder spurious interactions between two supercells, a vacuum buffer space of 10 Å is inserted in the z direction which is perpendicular to the plane of monolayer and heterolayers. All the cell parameters and the atomic coordinates are relaxed until the maximum Hellmann-

System	a (Å)	d_{Mo-S}/d_{W-S} (Å)	$\mathrm{d}_{B\text{-}N}(\mathrm{\mathring{A}})$	E_b (meV/atom)	$\mathbf{h}_{Mo\text{-}Mo}/\mathbf{h}_{W\text{-}W}(\text{\AA})$	$E_g (eV)$
Monolayer MoS ₂	3.185	2.418/-				1.850
Monolayer WS ₂	3.173	-/2.410				1.940
Monolayer h-BN	2.510		1.452			4.480
Bilayer MoS ₂	3.185	2.418/-		59.333	6.064/-	1.142 (indirect)
Bilayer WS ₂	3.173	-/2.410		59.833	-/5.845	1.051 (indirect)
Bulk MoS ₂	3.185	2.418/-			6.247/-	0.975 (indirect)
Bulk WS ₂	3.173	-/2.410			-/6.247	0.895 (indirect)

TABLE I. The calculated structural parameters such as a, lattice constant; d_{Mo-S} , d_{W-S} , and d_{B-N} , Mo-S, W-S, and B-N bond lengths; E_b , binding energies; h_{Mo-Mo} and h_{W-W} , vertical distance of two Mo atoms and two W atoms and E_g , computed band gap values.

Feynman force acting on each atoms is smaller than $0.03 \text{ eV}\text{\AA}^{-1}$. The total energy convergence is set to less than 10^{-6} eV in our calculations. The equilibrium geometries are fully optimised. To populate the electron state in the self-consistent calculations, the Methfessel Paxton scheme⁴⁶ is used with a smearing width of 0.05 eV.

III. RESULTS AND DISCUSSION

First, the lattice constants of single layer MoS₂, WS₂, and h-BN are calculated to be a = 3.185 Å, 3.173 Å, and 2.510 Å, respectively, as summarised in Table I. This is in reasonable agreement with previous results.^{47,48} Even though there is a quite difference between the lattice constants of MoS₂ or WS₂ and monolayer h-BN, all of them have a hexagonal structure, as shown in Figs. 1(a)-1(c). In order to set up an appropriate unit cell for MoS₂/h-BN and WS₂/ h-BN, and their sandwiched structures, we use 4×4 MoS₂ and WS₂ monolayers to match the 5×5 h-BN monolayer which results in an almost perfect match with the lattice mismatches of $\sim 1.6\%$ and $\sim 1.1\%$ for MoS₂/h-BN and WS₂/ h-BN heterostructures, respectively. These lattice mismatches are considerably small compared with previously studied heterostructures.^{49,50} We calculate Mo-S, W-S, and B-N bond lengths as 2.418 Å, 2.410 Å, and 1.452 Å, respectively, which are consistent with the reported values in Refs. 47, 51, and 52.

Next, we construct MoS_2 and WS_2 bilayer and WS_2/MoS_2 heterolayer systems as shown in Figs. 2(a)–2(c). For the bilayer and heterolayer systems, two different stacking orders, AA and AB, are considered. In AB stacking, the S atoms in the top layer are directly aligned above the Mo (W) atoms in the bottom layer, while AA stacking refers to Mo or W (S) atoms on the top layer right above the same atoms of the adjacent layer. Our test calculations suggest that AB stacking is the most energetically stable configuration. For the WS₂/MoS₂ heterostructure, we simply use the primitive cells of MoS₂ and WS₂ with negligible strain since the lattice mismatch between them is much smaller (0.5%) than MoS₂/ h-BN. The vertical distance between two Mo atoms, two W atoms, and Mo and W atoms in adjacent layers is found to be 6.064 Å, 5.845 Å, and 6.148 Å for bilayer MoS₂, bilayer WS₂, and WS₂/MoS₂ heterolayer, respectively, which indicates a van der Waals type interaction between the layers. These distances have excellent agreement with reported distances in Ref. 53. Moreover, the similar interlayer distances between two Mo and two W atoms in adjacent layers in bulk MoS_2 and WS_2 are calculated as 6.247 Å, which are in very good agreement with the experimental value [6.220 Å (Refs. 53 and 54)], hence the vdW-corrected functional is reliable for these calculations. We calculate the total energy difference between the AA and AB stacking orders for the WS₂/ MoS₂ heterolayer; the AB stacking is energetically favored with an energy difference of 0.069 eV/cell. The binding energies (E_b) between the layers for bilayer MoS₂ and WS₂, and heterolayer WS_2/MoS_2 are defined as

$$E_b = [E(MoS_2 \, bilayer) - 2 \times E(MoS_2 \, monolayer)]/N, \quad (1)$$

$$E_b = [E(WS_2 \, bilayer) - 2 \times E(WS_2 \, monolayer)]/N, \quad (2)$$

$$E_b = [E(WS_2/MoS_2 heterobilayer) - E(MoS_2 monolayer) - E(WS_2 monolayer)]/N,$$
(3)

where $E(MoS_2 \text{ bilayer})$, $E(WS_2 \text{ bilayer})$, and $E(WS_2/MoS_2 \text{ heterobilayer})$ are the total energies of the $MoS_2 \text{ bilayer}$, $WS_2 \text{ bilayer}$, and WS_2/MoS_2 heterobilayer, respectively; E_{MoS_2} and E_{WS_2} are the total energies of MoS_2 and WS_2 monolayers, respectively; and *N* is the total number of atoms in the supercell. The binding energies per atom are listed in Tables I and II. We found almost no differences between binding energies for bilayer MoS_2 and WS_2 . However, when the MoS_2 monolayer is stacked on the WS_2 monolayer, the binding energy is increased by an amount of 10 meV, making the interlayer interaction slightly stronger. We further found a good agreement with available values reported in Refs. 54 and 55. This is another indication that the interlayer bonding of MoS_2 and WS_2 is weak van der Waals interactions. We

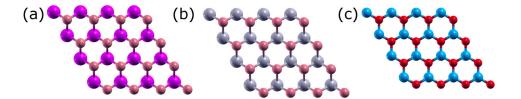
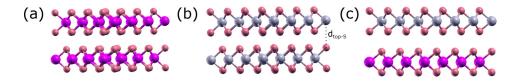
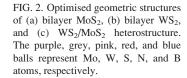


FIG. 1. Optimised geometric structures of (a) monolayer MoS_2 , (b) monolayer WS_2 , and (c) monolayer h-BN. The purple, grey, pink, red, and blue balls represent Mo, W, S, N, and B atoms, respectively.



also notice that there are no significant changes in Mo-S and W-S bond lengths when the lattice constant of either MoS_2 or WS_2 is used in the WS_2/MoS_2 heterostructure.

To obtain the equilibrium geometry of MoS₂/h-BN, WS₂/h-BN, and WS₂/MoS₂/h-BN heterostructures and their sandwiched structures, we optimised the heterostructures using several starting positions of MoS₂, WS₂, and WS₂/ MoS₂ subject to the h-BN one. After optimisation calculations, we found the most stable configuration with one Mo (W) atom on top of the B atom and the S atom near the N atom site on the bottom layer of the heterostructures, as shown in Figs. 3(a)-3(f). For sandwiched structures, different stacking orders are considered by rotating the top monolayer h-BN. After our total energy calculations, we found the total energy differences being less than 9.685 meV between different types of their stacking orders. Therefore, the ABA stacking is used in all sandwiched structures. The structural parameters are summarised in Table II for the heterostructures. The interlayer spacing between monolayer h-BN and the topmost S atom in MoS₂ and WS₂ is estimated as 3.451 Å and 3.425 Å for MoS₂/h-BN and WS₂/h-BN heterostructures, respectively. For MoS₂/h-BN heterostructures, the Mo-S and B-N bond lengths are found to be 2.420 Å and 1.470 Å, respectively, which represent 1.4% stretching of the monolayer h-BN substrate. The substrate is also slightly buckled with the amount of 0.18 Å. In the WS₂/h-BN heterostructure, the vertical distance of the topmost S atoms in WS_2 from monolayer h-BN is found to be 3.425 Å together with B-N and W-S bond lengths of 1.470 Å and 2.410 Å, respectively. Interestingly, while the monolayer h-BN is buckled in the MoS₂/h-BN heterostructure, the monolayer h-BN is almost flat with the buckling of 0.03 Å in WS₂/h-BN. When the WS_2/MoS_2 heterostructure is deposited on monolayer h-BN, the interfacial distance between MoS₂ and WS₂ monolayers is increased from 6.148 Å to 6.303 Å representing the weakening the interaction between the layers. The interlayer spacing between monolayer h-BN and the topmost S atoms in MoS_2 is found to be 3.512 Å with the negligible rumpling of 0.01 Å in the h-BN monolayer. The



computed binding energies for MoS₂/h-BN, WS₂/h-BN, and WS₂/MoS₂/h-BN heterostructures are listed in Table II. The binding energy of these heterostructures is defined as

$$E_{b} = \left[E_{MoS_{2}/h-BN} - E_{MoS_{2}} - E_{h-BN} \right] / N, \qquad (4)$$

$$E_{b} = \left[E_{WS_{2}/h-BN} - E_{WS_{2}} - E_{h-BN} \right] / N, \tag{5}$$

$$E_b = \left[E_{WS_2/MoS_2/h-BN} - E_{MoS_2} - E_{WS_2} - E_{h-BN} \right] / N, \quad (6)$$

where $E_{MoS_2/h-BN}$, $E_{WS_2/h-BN}$, and $E_{WS_2/MoS_2/h-BN}$ are the total energies of the MoS2/h-BN, WS2/h-BN, and WS2/ MoS_2/h -BN heterostructures, respectively; E_{MoS_2} , E_{WS_2} , and E_{h-BN} are the total energies of MoS₂, WS₂, and h-BN monolayers, respectively; and N is the total number of atoms in the heterostructure. For sandwiched structures, the calculated equilibrium interlayer distances are 3.566 Å, 3.583 Å, and 3.476 Å for h-BN/MoS₂/h-BN, h-BN/WS₂/h-BN, and h-BN/ WS₂/MoS₂/h-BN systems, respectively. However, for the h-BN/WS₂/MoS₂/h-BN system, we found that the interlayer distance between the top h-BN monolayer and the topmost S atoms in the WS₂ monolayer is 3.476 Å, which is smaller than that of the distance between the bottom h-BN monolayer and the topmost S atoms in the MoS₂ monolayer. For all sandwiched structures, a negligible buckling of 0.02 Å in both top and bottom layers of h-BN is induced by the MoS₂, WS₂, and WS₂/MoS₂ layers. To analyse the stabilities of these sandwiched heterostructures, the binding energy of the heterostructure is defined with the following equations:

$$E_b = \left[E_{h-BN/MoS_2/h-BN} - E_{MoS_2} - 2 \times E_{h-BN} \right] / N, \quad (7)$$

$$E_{b} = \left[E_{h-BN/WS_{2}/h-BN} - E_{WS_{2}} - 2 \times E_{h-BN} \right] / N, \quad (8)$$

$$E_{b} = \left[E_{h-BN/WS_{2}/MoS_{2}/h-BN} - E_{MoS_{2}} - E_{WS_{2}} - 2 \times E_{h-BN} \right] / N,$$
(9)

where $E_{h-BN/MoS_2/h-BN}$, $E_{h-BN/WS_2/h-BN}$, and $E_{h-BN/WS_2/h-BN}$ are the total energies of the h-BN/MoS_2/h-BN, h-BN/WS_2/h-BN, and h-BN/WS_2/h-BN

TABLE II. The calculated structural parameters such as a, lattice constant; d_{Mo-S} , d_{W-S} , and d_{B-N} , Mo-S, W-S, and B-N bond lengths; Δ_{BN} , buckling of the monolayer h-BN; E_b , binding energies; h_0 , interlayer spacing between monolayer h-BN and the topmost S atom in MoS₂ and WS₂; h_{Mo-W} , vertical distance between Mo and W atoms and E_g , computed band gap values.

System	$\mathrm{d}_{Mo\text{-}S}/\mathrm{d}_{W\text{-}S}(\mathrm{\mathring{A}})$	d_{B-N}/Δ_{BN} (Å)	E_b (meV/atom)	h_0/h_{Mo-W}	$E_g (eV)$
WS ₂ /MoS ₂	2.418/2.410		49.833	-/6.148	0.907 (indirect)
MoS ₂ /h-BN	2.420/-	1.470/0.18	41.225	3.451/-	1.710 (indirect)
WS ₂ /h-BN	-/2.410	1.470/0.03	42.531	3.425/-	1.920
WS ₂ /MoS ₂ /h-BN	2.418/2.410	1.460/0.01	46.322	3.512/6.303	0.967 (indirect)
h-BN/MoS ₂ /h-BN	2.417/-	1.471/0.02	53.101	3.566/-	0.960 (indirect)
h-BN/WS ₂ /h-BN	-/2.410	1.465/0.02	54.081	3.583/-	1.230 (indirect)
h-BN/WS ₂ /MoS ₂ /h-BN	2.415/2.410	1.471/0.03	55.219	3.476/6.171	1.110 (indirect)

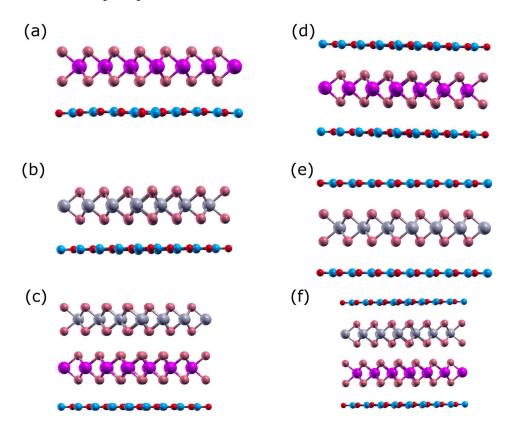


FIG. 3. Optimised geometric structures of (a) MoS_2/h -BN, (b)WS_2/h-BN, (c) WS_2/MoS_2/h-BN, (d) h-BN/MoS_2/h-BN h-BN, (e) h-BN/WS_2/h-BN, and (f) h-BN/WS_2/MoS_2/h-BN heterostructures. The purple, grey, pink, red, and blue balls represent Mo, W, S, N, and B atoms, respectively.

heterostructures, respectively; E_{MoS_2} , E_{WS_2} , and E_{h-BN} are the total energies of MoS₂, WS₂, and h-BN monolayer, respectively; and N is the total number of atoms in the heterostructure. We obtained a negative binding energy for all heterostructures suggesting that the formation of heterostructures is an exothermic process from the thermodynamic point of view. This is also an indication that these heterostructures are energetically favorable. As summarised in Table II, the smaller binding energy represents the strongest binding between the layers in the heterostructures. The values of E_{h} are in the range of physical adsorption which suggests weak vdW interactions between the layers. Our results also suggest that the interlayer interaction in MoS₂/h-BN, WS₂/h-BN, and WS₂/MoS₂/h-BN heterostructures is slightly stronger than that in the sandwiched structures. From the present results, we suggest that MoS₂, WS₂, and WS₂/MoS₂ layers interact very weakly with the h-BN monolayer due to the large distance between the TMD layers and the h-BN monolayer, and a small rumpling in h-BN. This also leads to significantly weak electronic coupling between the layers and the h-BN monolayer.

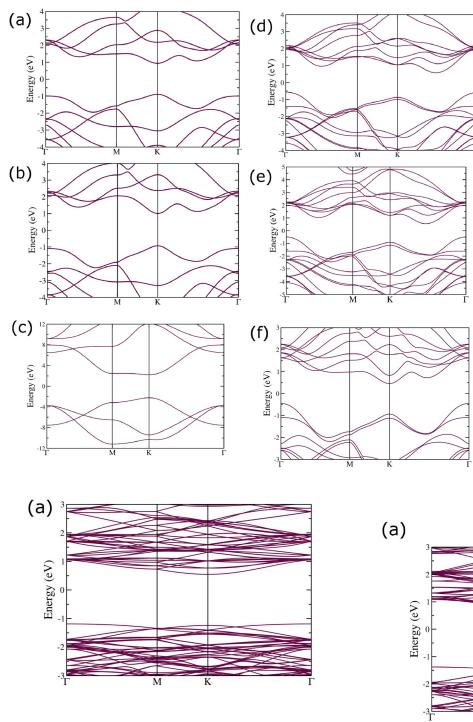
After the determination of the structural stability of h-BN/MoS₂/h-BN, h-BN/WS₂/h-BN, and h-BN/WS₂/MoS₂/h-BN heterostructures, we now present the electronic band structures of these heterostructures. Figures 4(a)–4(c) shows the electronic structure of monolayer MoS₂, WS₂, and h-BN. Our vdW-DF calculations indicate that both monolayer MoS₂ and monolayer WS₂ have direct band gaps of 1.850 eV and 1.940 eV, respectively, consistent with the previously reported studies.^{20,24,49} As shown in Fig. 4(c), we found a large band gap of 4.480 eV for monolayer h-BN with vdW-DF calculations, closer to the experimental value.⁵⁶ Because of this large band gap, monolayer h-BN is considered as the best candidate for enhancing the band gap in vdW heterostructures. For the WS₂/MoS₂ heterostructure, the indirect band gap of 0.907 eV is obtained as shown in Fig. 4(f). This indirect band gap is due to the interlayer coupling between monolayer MoS₂ and monolayer WS₂. The band gap is also considerably smaller than that of both monolayer MoS₂ and WS₂. Similarly, the in-direct band gaps of 0.975 eV and 0.895 eV are observed in bulk MoS₂ and bulk WS₂, respectively, in good agreement with experimental values.^{57,58} When a second layer of MoS₂ and WS₂ is put on the monolayer MoS₂ and WS₂, the direct band gap is transformed into the indirect band gap. As presented in Figs. 4(d) and 4(e), our calculated gap values are 1.142 eV and 1.051 eV for bilayer MoS₂ and WS₂, respectively.

After depositing MoS₂ on monolayer h-BN, the direct band gap nature of MoS₂ changes to an indirect band gap. As shown in Fig. 5(a), the calculated direct gap at the K point and the indirect gap are 1.805 eV and 1.710 eV, respectively. When another monolayer h-BN is added on top of MoS₂/h-BN heterostructure, we notice that the indirect band gap is preserved with the value of 0.960 eV, as presented in Fig. 5(b). However, we found that the direct band gap at the **K** point is decreased to 1.112 eV, and the indirect band gap value is also reduced to 0.960 eV. In both interfaced and sandwiched structures, the valence band maximum (VBM) at the **K** point is smaller than the valence band edge at the Γ point. When the WS₂ layer is stacked on the h-BN monolayer, the direct band gap remained with the value of 1.920 eV, as indicated in Fig. 6(a). However, the direct band gap is transformed into an indirect band gap with the value of 1.230 eV [shown in Fig. 6(b)], when the WS₂ monolayer is intercalated into bilayer h-BN. It is worth mentioning that the direct band gap value of WS₂ is slightly decreased with

FIG. 4. Band structures of monolayer (a) MoS_2 , (b) WS_2 , (c) h-BN, (d) bilayer

 MoS_2 , (e) bilayer WS_2 , and (f) WS_2 / MoS_2 heterolayer. The Fermi level is set

to zero.



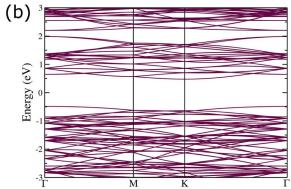


FIG. 5. Band structures of (a) MoS_2/h -BN and (b) h-BN/MoS_2/h-BN heterostructures. The Fermi level is set to zero.

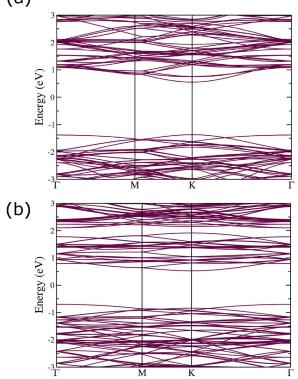


FIG. 6. Band structures of (a) WS_2/h -BN and (b) h-BN/WS_2/h-BN hetero-structures. The Fermi level is set to zero.

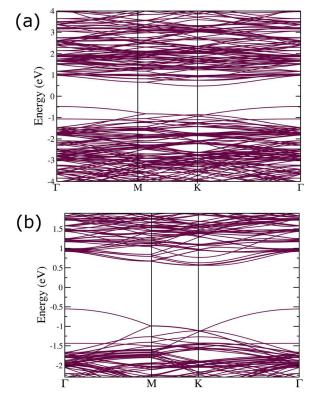


FIG. 7. Band structures of (a) $WS_2/MoS_2/h$ -BN and (b) h-BN/WS_2/MoS_2/h-BN heterostructures. The Fermi level is set to zero.

the amount of 20 meV when compared with the WS₂ heterostructure, exhibiting no noticeable covalent bonding of the layers. The calculated band structures of WS₂/MoS₂/h-BN and h-BN/WS₂/MoS₂/h-BN heterostructures are presented in Figs. 7(a) and 7(b). They are both indirect band gap semiconductors with the value of 0.967 eV and 1.110 eV, respectively. It is clear that the indirect band gap of WS₂/MoS₂ is reduced to 0.967 eV when the WS₂/MoS₂ heterostructure is deposited on monolayer h-BN. However, the indirect band gap for the WS₂/MoS₂ heterostructure varies from 0.907 eV to 1.110 eV when inserting the WS₂/MoS₂ heterostructure into bilayer h-BN implying the protection of its intrinsic properties.

To further analyse the electronic interaction between monolayer MoS₂, WS₂, and h-BN, we calculate the total density of states (TDOS) and partial density of states (PDOS) including the d-electrons of the transition metals and the p-electrons of chalcogen, boron, and nitrogen atoms. Figures 8(a) and 8(b) show the TDOS and the PDOS of the WS₂/MoS₂ heterostructure. The valence band maximum (VBM) is contributed by d orbitals of both W and Mo atoms and the p orbital of the S atom, while the conduction band minimum (CBM) is mainly dominated by the d orbitals of Mo atoms. For the MoS₂/h-BN system, PDOS analysis reveals that the states below the Fermi level are slightly dominated by only p orbitals of N atoms [see Figs. 9(a) and 9(c)], while there is almost no contribution from h-BN monolayer above the Fermi level. As clearly shown in Fig. 9(c), most states in both occupied and unoccupied regions arose from Mo d-electrons for MoS₂/h-BN. Intercalating a monolayer MoS₂ into a bilayer h-BN increases the negligible contribution of the p orbital of B atoms above the Fermi level, while vanishing the contribution of p orbitals of N atoms below the Fermi level as presented in Figs. 9(b) and 9(d). Additionally, new peaks are generated at E_t +0.725 eV in the unoccupied region, at E_{f} -0.725 eV and E_{f} -1.115 eV in the occupied region, which are mostly dominated by the d electrons of Mo atoms. Furthermore, both peaks at E_f+1.051 eV and E_f-2.015 eV in MoS₂/h-BN structures are shifted when another h-BN monolayer is added to the MoS₂/ h-BN heterolayer. The PDOS and TDOS for WS₂/h-BN and h-BN/WS₂/h-BN heterostructures are shown in Figs. 10(a)-10(d). The conduction band states are mainly due to the d orbitals of W atoms for both heterostructures. However, we notice that there is a significant contribution from the p orbitals of the B atoms at $E_f+2.8 \text{ eV}$ in the WS₂/ h-BN heterostructure. The states on the valance bands are generally due to hybridisation of the p orbitals of S and N atoms and d orbitals of W atoms in the WS₂/h-BN heterostructure. Furthermore, the contribution of p orbitals of B atoms in conduction band states is significantly reduced when WS₂ was sandwiched into the h-BN bilayer. Similarly, the contribution of p orbitals of N atoms disappeared in valence band states resulting only in hybridisation between p orbitals on S atoms and d orbitals on W atoms. The PDOS of $WS_2/MoS_2/h-BN$ [see Figs. 11(a)-11(d)] shows that the VBM is mostly dominated by d orbitals of Mo and W atoms, while the CBM is mainly dominated by only d orbitals of Mo atoms. Moreover, there is a small contribution of p orbitals of B and N atoms to states in conduction bands and valence bands, respectively. The change in the PDOS when another h-BN monolayer is added on top of the WS₂/MoS₂/ h-BN heterostructure is that the states in the VBM are due to the d orbitals of metal atoms and the p orbitals of S atoms in the MoS_2 layer and those of N atoms, as shown in Fig. 11(d). The states in the CBM are due to the d orbitals of Mo atoms and p orbitals of S atoms in the MoS₂ layer, while there is an absence of contribution of p orbitals of S atoms in the $WS_2/$

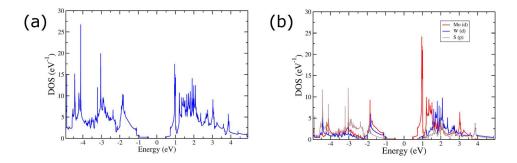


FIG. 8. Total density of states (TDOS) for (a) WS_2/MoS_2 and partial density of states (PDOS) for (b) WS_2/MoS_2 heterostructures. The Fermi level is at zero energy.

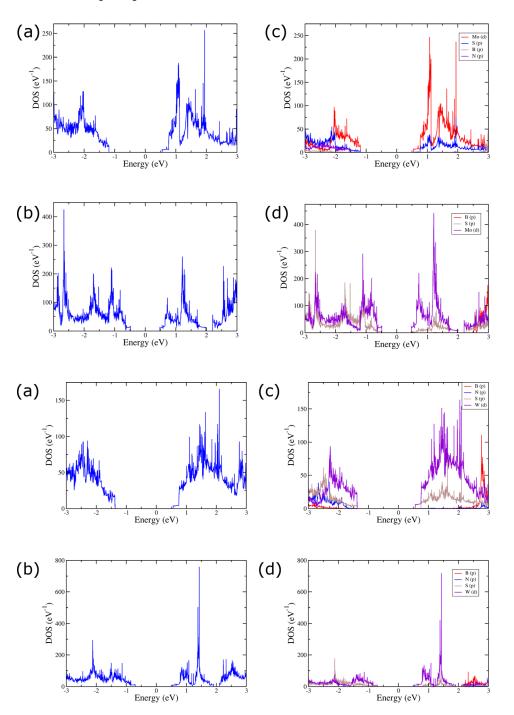


FIG. 9. Computed TDOS for (a) MoS_2/h -BN and (b) h-BN/MoS_2/h-BN and PDOS for (c) MoS_2/h -BN and (d) h-BN/MoS_2/h-BN. The Fermi level is at zero energy.

FIG. 10. Computed (TDOS) for (a) WS_2/h -BN and (b) h-BN/WS_2/h-BN and PDOS for (c) WS_2/h -BN and (d) h-BN/WS_2/h-BN heterostructures. The Fermi level is at zero energy.

MoS₂/h-BN heterostructure. We further notice that the contribution of p orbitals of B atoms in both top and bottom layers of h-BN is significantly increased at E_f +3.150 eV in the unoccupied region.

IV. CONCLUSION

In conclusion, the electronic properties and dynamic stability of van der Waals heterostructures of h-BN/MoS₂/h-BN, h-BN/WS₂/h-BN, and h-BN/WS₂/MoS₂/h-BN are investigated using density functional theory with a recently developed non-local van der Waals (vdW) density functional by Vydrov and Van Voorhis (rVV10). The calculated binding energies for different stacking orders confirm the dynamic stability of these heterostructures, and it was indicated that the van der Waals interaction between these two layers was not strong enough to make an abrupt structural change in TMDs. Therefore, the proposed heterostructures here offer a significant advantage to producing TMD based optoelectronic device applications. By analysing the electronic structures of the heterostructures, we found that all considered heterostructures are an indirect band gap semiconductor with the exception of the WS₂/h-BN heterostructure representing a direct band gap semiconductor. From band structure calculations, we conclude that the valence band maximum (VBM) at the Γ point is very sensitive to interlayer interaction in the heterostructures which defines the location of the VBM at the Γ or **K** point. Our PDOS calculations represent the negligible interaction between h-BN and TMDs. The states in the conduction band are also not affected by interlayer

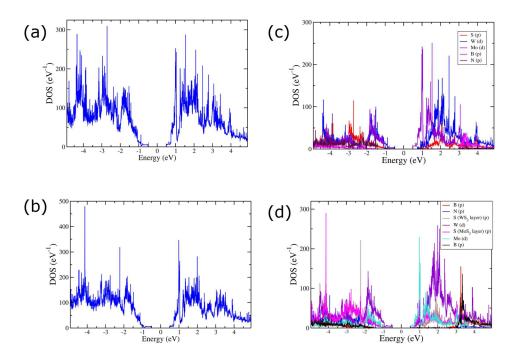


FIG. 11. Computed TDOS for (a) WS₂/MoS₂/h-BN and (b) h-BN/WS₂/ MoS₂/h-BN and PDOS for (c) WS₂/ MoS₂/h-BN and (d) h-BN/WS₂/MoS₂/ h-BN heterostructures. The Fermi level is at zero energy.

interactions. This suggests that h-BN monolayer can be used to protect monolayer TMDs, and the heterostructures studied here offer significant advantages to producing TMD based optoelectronic device applications. We predict that the $WS_2/$ h-BN heterostructure may be useful for solar energy conversion due to retaining its direct band gap. The direct band gap heterostructures should have higher efficiency in solar energy applications when compared with indirect band gap heterostructures. We believe that these results provide a support for experiments and serve as a guide for the development of novel 2D structures with high performance for photovoltaic and optoelectronic devices. More importantly, it is found that h-BN is an excellent candidate for the protection of intrinsic properties of MoS_2 , WS_2 , and WS_2/MoS_2 structures.

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